

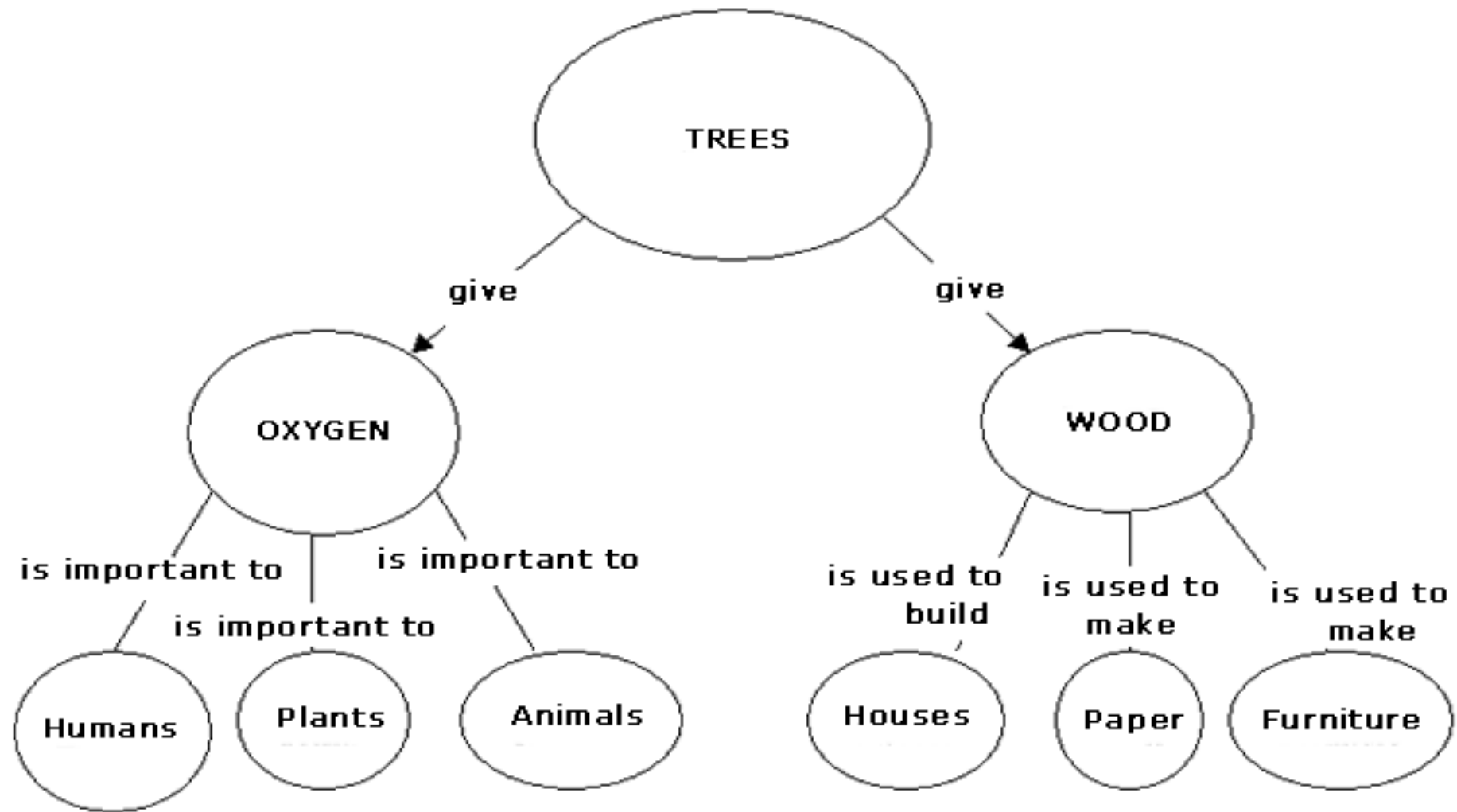
**GULU University February 2008**

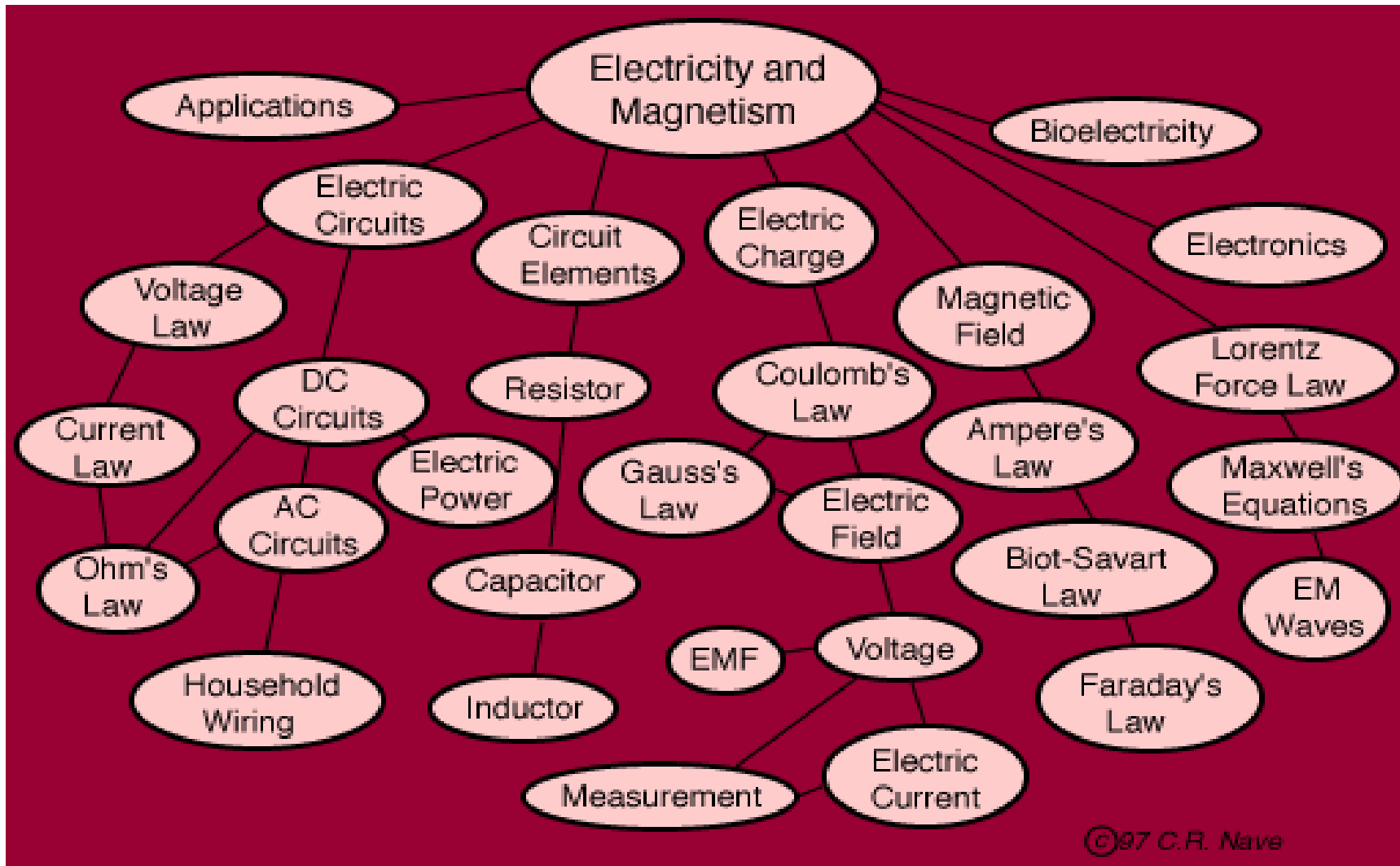
**Course 105 ELECTRICITY and MAGNETISM**

**prof. Elena Sassi University of Naples “Federico II” Italy**

**sassi@na.infn.it**

## Concept mapping





## Some “fathers” of Electromagnetism



H. Christian Ørsted  
(1777 - 1851)



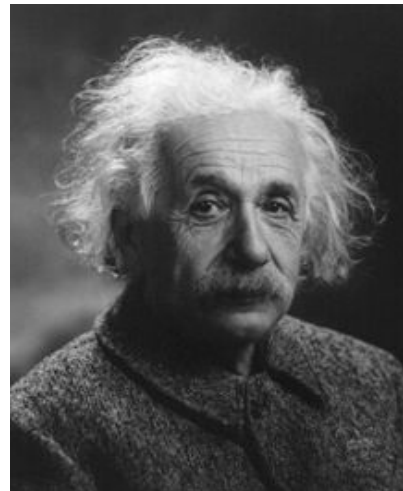
André-Marie Ampère  
(1775-1836)



Michael Faraday  
(1791-1867)



James Clerk Maxwell  
(1831 -1879)



Albert Einstein  
(1879 -1955)

# ELECTRICITY?



Ambiguous term from Physics viewpoint

ELECTRIC FIELD

ELECTRIC POTENTIAL

ELECTRIC CHARGE

ELECTRIC CURRENT

ELECTRIC ENERGY

ELECTRIC POWER

## Electrostatics

phenomena arising from stationary **electric charges**

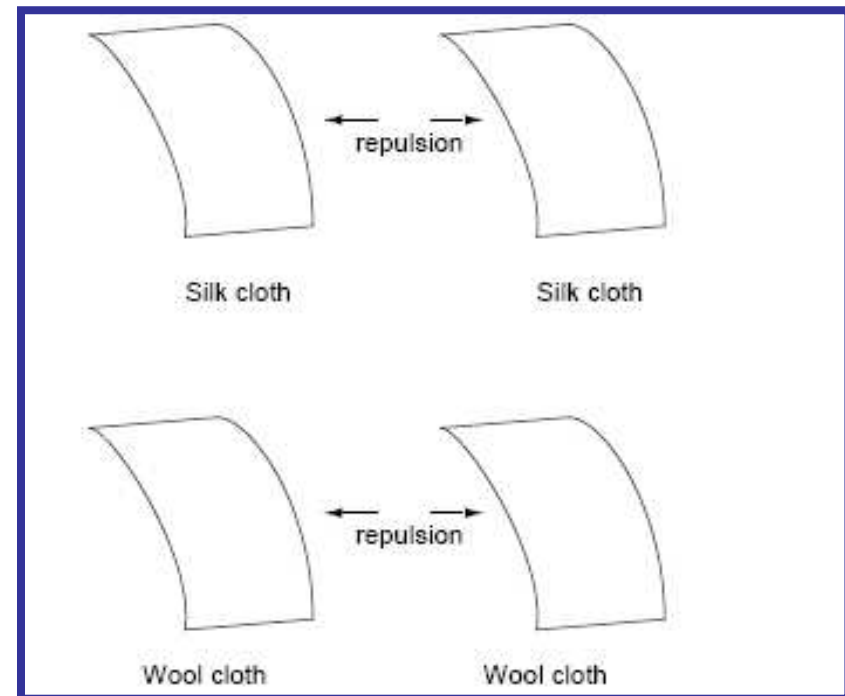
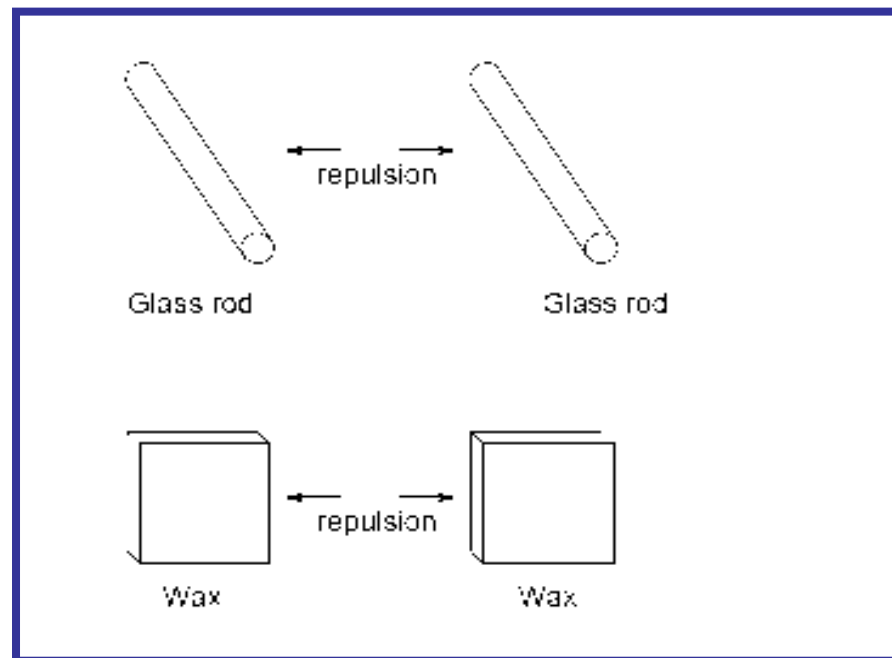
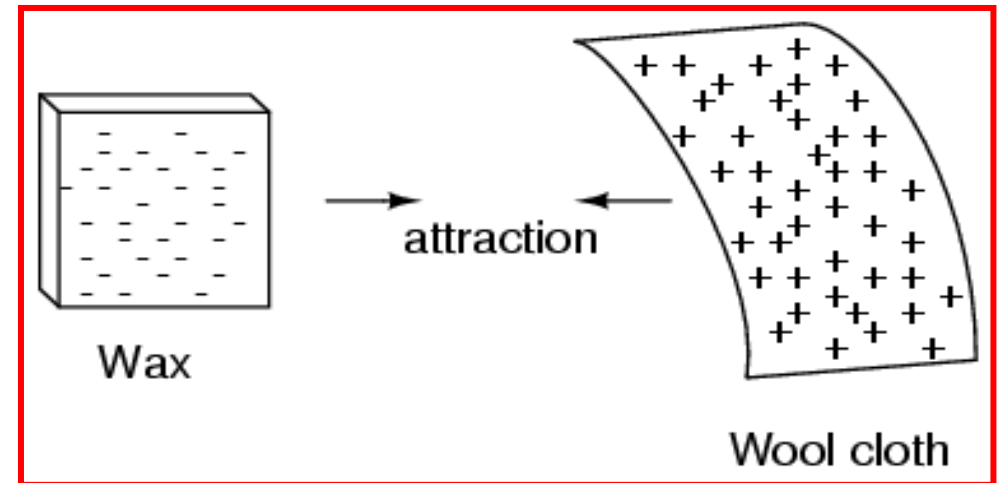
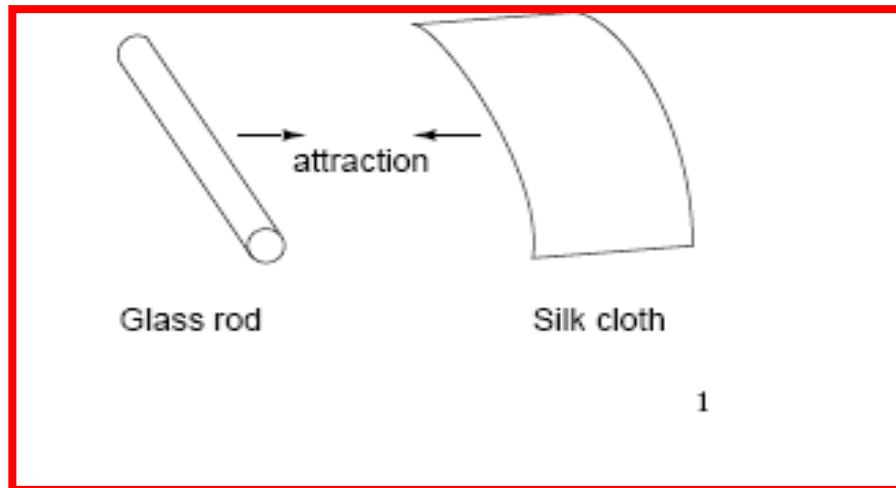
Some key ideas for Teaching Electrostatic

- Electrostatic in everyday life
- Attraction, repulsion, sparks
- Rubbing objects with wool cloth, human and animal hair
- Charge induction
- Imbalanced number of electrons and protons in an object
- Microscopic viewpoint
- .....



amber

## Electrostatic phenomena: low cost experiments



Experimental demonstrations *best performed with very little relative humidity*

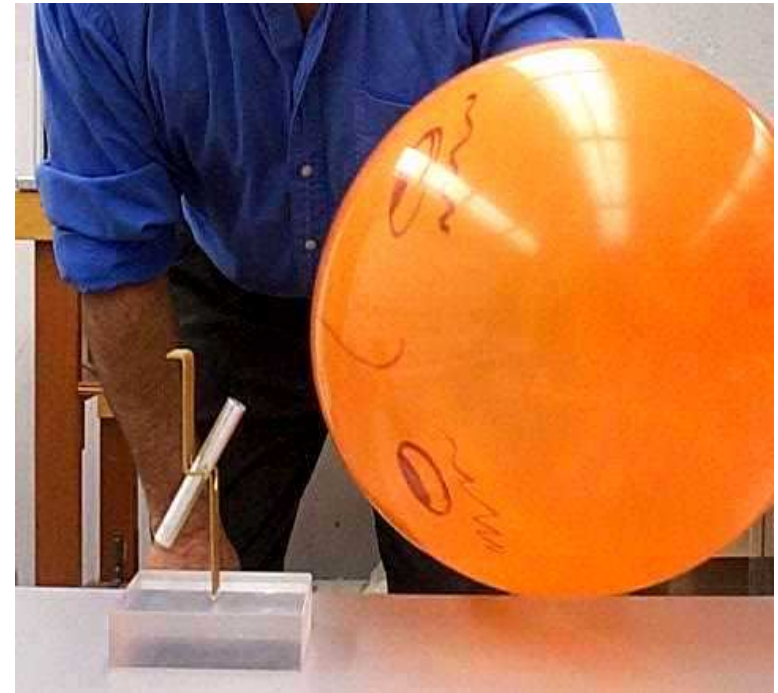
- 1) Jumping paper: small paper pieces attracted by a rubbed plastic pen
- 2) A balloon, rubbed on human hair, sticks on a wall
- 3) A rolling can: a soft drink can on its side is on a smooth surface. A large rubbed balloon is brought near the can that can roll towards it



## ELECTROSTATIC DEMONSTRATIONS

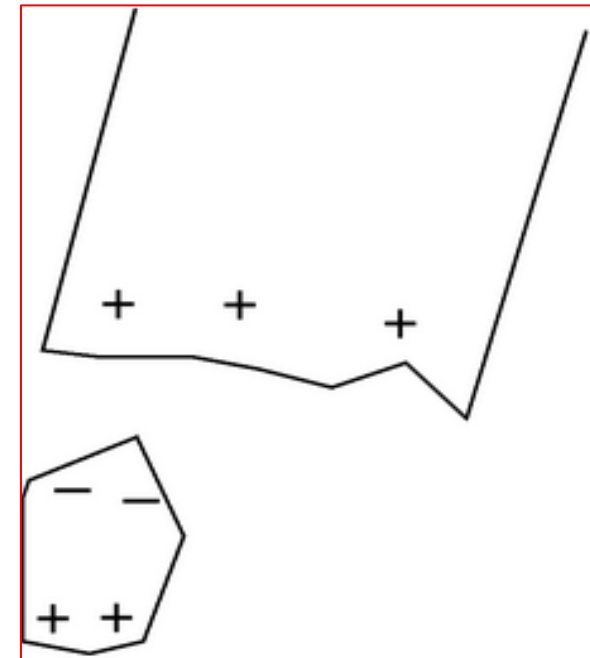
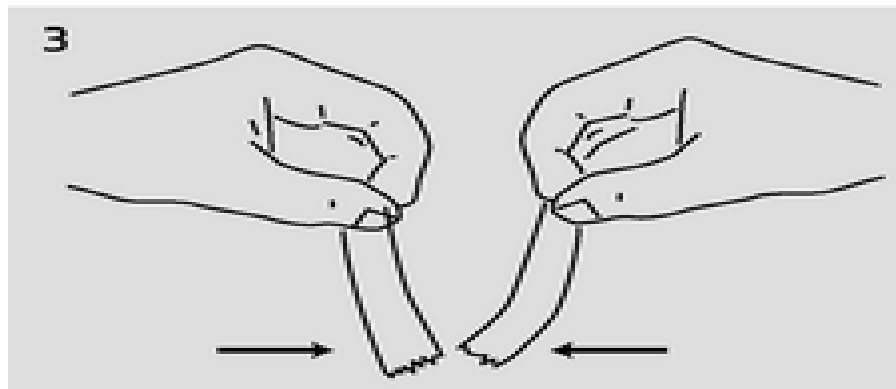
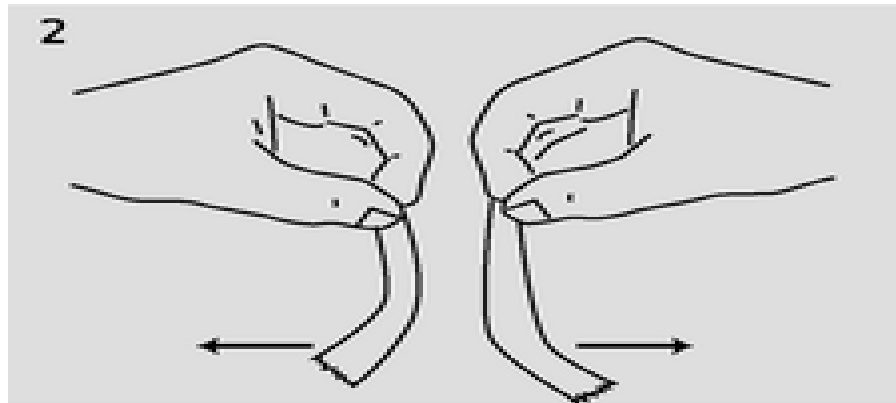
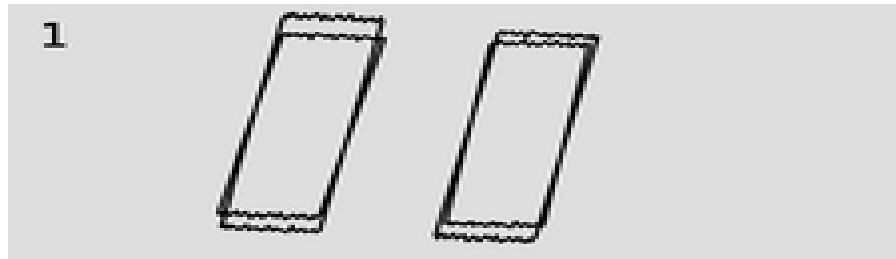
4) CHARGE METER      A bent piece of brass is mounted on a plastic or wood base. An Al tube is placed in the appropriate slot in the brass. When charge is near, the Al tube bends towards the charged object

5) THE ADMIRER BALLOON Draw a face on an inflated balloon and suspend it at the level of your head. Rub the face of the balloon; it will move toward you whenever you approach it.



## Tribo effect

4 Pieces of tape



## TRIBOELECTRIC SERIES

***Most positively charged +***

**Human skin**

**Leather**

**Rabbit's fur**

**Glass**

**Quartz**

**Mica**

**Human hair**

**Nylon**

**Wool**

**Cat's fur**

**Silk**

**Aluminium**

**Paper (*Small + charge*)**

**Cotton (*No charge*)**

**Wood (*Small charge*)**

**Amber**

**Sealing wax**

**Rubber balloon**

**Resins**

**Hard rubber**

**Rayon**

**Synthetic rubber**

**Polyester**

**Styrofoam**

**Plastic wrap**

**Polyethylene (like Scotch tape)**

**PolypropyleneVinyl (PVC)**

**Silicon**

**Teflon**

**Ebonite**

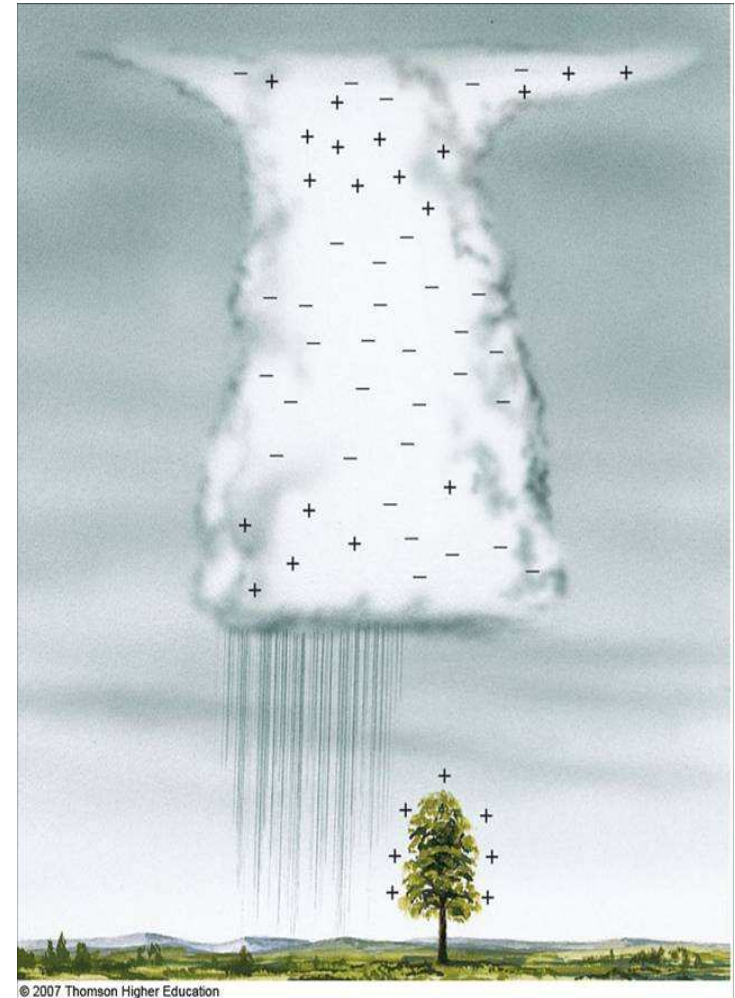
***Most negatively charged -***

**How many types of charge?**

**Positive and negative**

Point-like Electric charge: an idealized model  
of a charged particle  
It is an electric charge at a mathematical point  
with no dimensions

Approximation  
the dimension of the charged objects are  
much smaller than the other distances  
involved



Electron mass =  $9.1095 \times 10^{-31}$  kg

Electron charge =  $1.602 \times 10^{-19}$  coulomb

Electron radius =  $2.8179 \times 10^{-13}$  cm<sup>12</sup>

How do point-like charges interact?

$$F = k q_1 q_2 (1/r^2) r$$

**F** = electrostatic force, **k** = Coulomb's constant

**q<sub>1</sub>, q<sub>2</sub>** = first and second charge

**r** = distance between the two charges

**F** is a vector, like all forces

**Direction** = segment connecting the charges

**F** = Attractive if charges of same sign

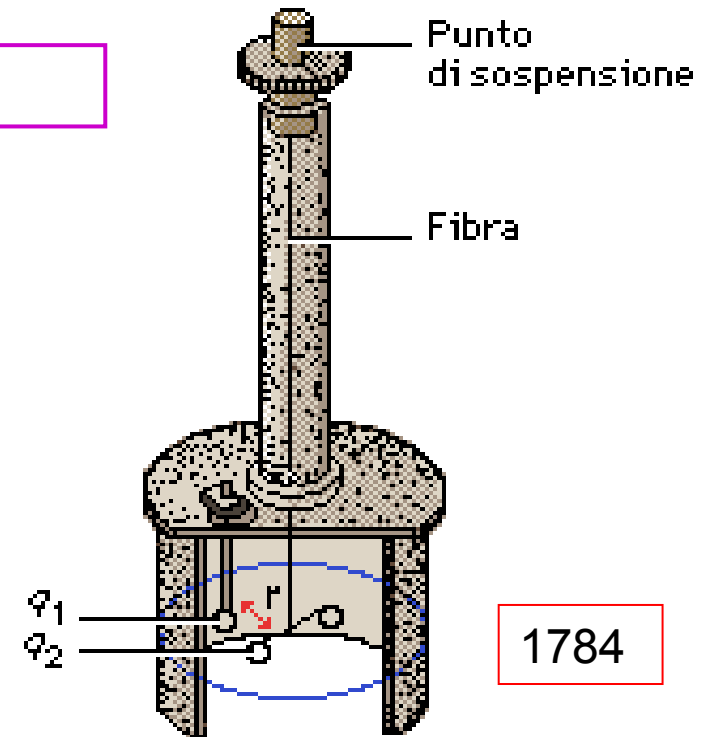
**F** = Repulsive if charges of opposite sign

**Dimension of a Force** =  $\text{m l t}^{-2}$

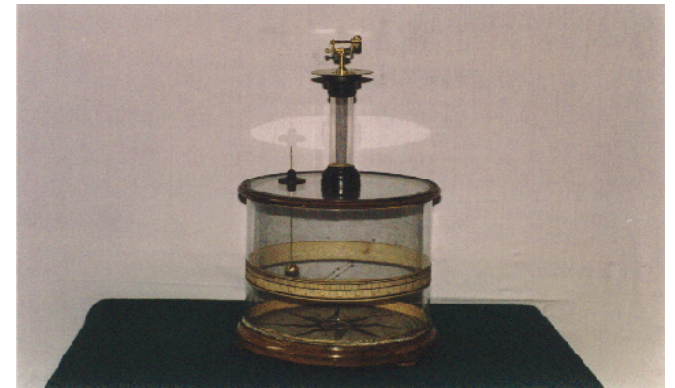
**Unit = Netwon** =  $\text{kg meter second}^{-2}$

Unit of charge = Coulomb =

$6.24 \times 10^{18}$  charge of an electron

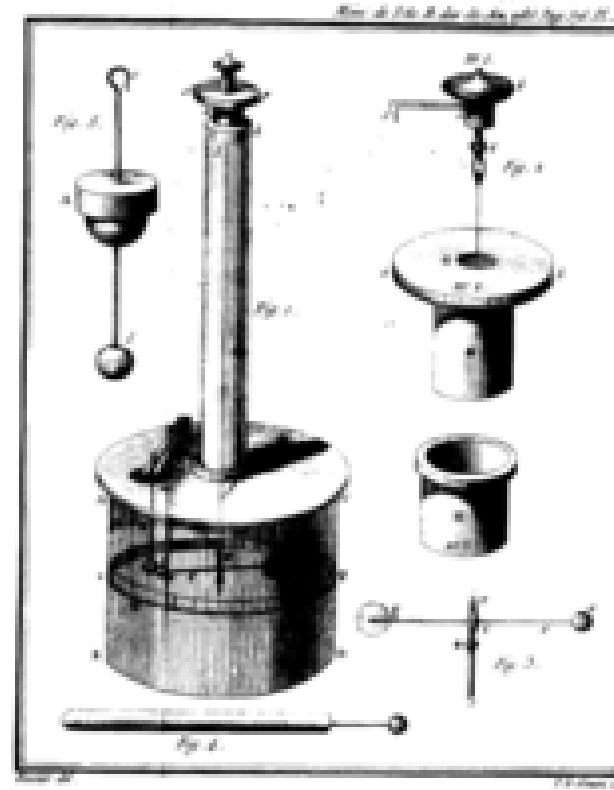
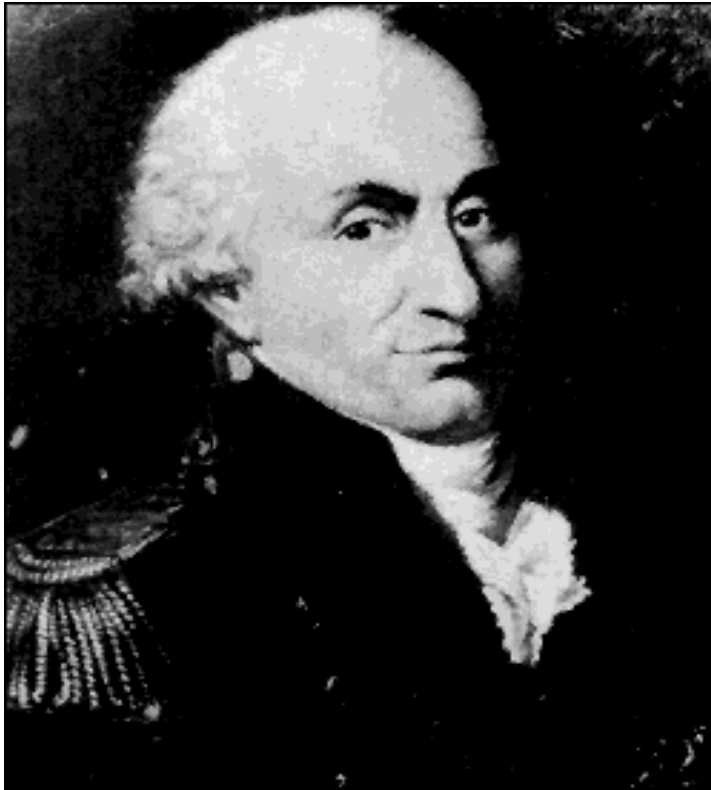


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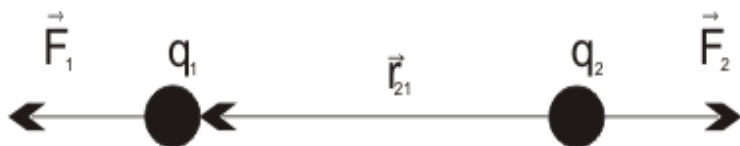


**Torsion Balance**

## Electrostatic Force: Coulomb's Law



$$\mathbf{F} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2 (\mathbf{r}_1 - \mathbf{r}_2)}{|\mathbf{r}_1 - \mathbf{r}_2|^3} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \hat{\mathbf{r}}_{21}$$



$$k_C = \frac{1}{4\pi\epsilon_0} \approx 8.988 \times 10^9$$

$$\text{N m}^2 \text{C}^{-2}$$

## Parallels between electrostatics and gravity

$$\mathbf{F} = \frac{1}{4\pi\epsilon_0} \frac{Qq}{r^2} \hat{\mathbf{r}} = q\mathbf{E}$$

$$\mathbf{F} = G \frac{Mm}{r^2} \hat{\mathbf{r}} = m\mathbf{g}$$

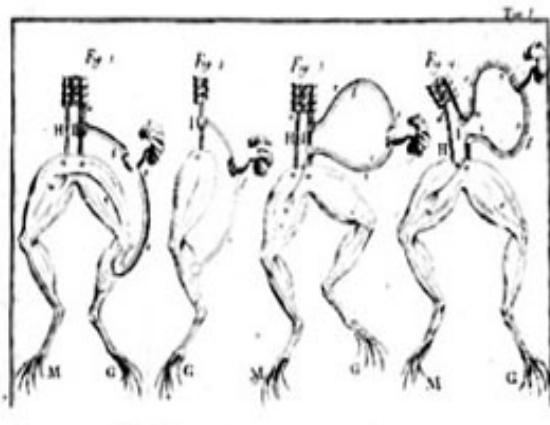
Central and Conservative forces

Inverse-square dependence from distance

No negative mass



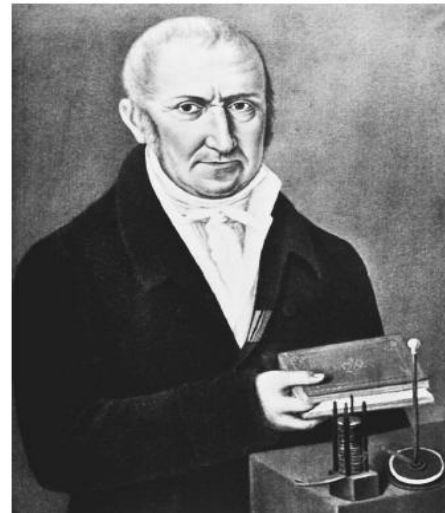
## Some “fathers” of Electricity



Boyle R.



Franklin B.



Volta A.

Faraday M.

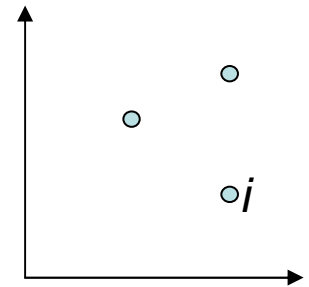


Galvani L.



What happens when there are N point-like charges?

$$\mathbf{F}(\mathbf{r}) = \frac{q}{4\pi\epsilon_0} \sum_{i=1}^N \frac{q_i(\mathbf{r} - \mathbf{r}_i)}{|\mathbf{r} - \mathbf{r}_i|^3} = \frac{q}{4\pi\epsilon_0} \sum_{i=1}^N \frac{q_i}{R_i^2} \hat{\mathbf{R}}_i$$



$q_i$  and  $\mathbf{r}_i$  = magnitude and position of the  $i_{th}$  charge

$\hat{\mathbf{R}}_i$  is a unit vector pointing from charge  $q_i$  to the test charge  $q$

$\mathbf{R}_i = \mathbf{r} - \mathbf{r}_i$  its magnitude  $R_i$  is the distance between charges  $q_i$  and  $q$

Linear Superposition

